

A Case of Disparity Between Pulse Oximetry Measurements and Blood Gas Analysis Values

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Case Summary

A 79-year-old man was admitted to the intensive care unit because of shortness of breath, dizziness, and confusion. He had a history of T cell prolymphocytic leukemia, which was not responding to therapy with fludarabine. His admission serum leukocyte count was 886,000 cells/ μ L and his vital signs were: temperature 36.9°C, respiratory rate 28 breaths/min, heart rate 91 beats/min, and blood pressure 160/91 mm Hg. Chest auscultation revealed bibasilar rales. A chest computed tomogram (CT) revealed a diffuse ground-glass appearance of the lung fields and no evidence of pulmonary emboli. While receiving supplemental oxygen via a nonrebreather mask, his arterial oxygen saturation (measured via pulse oximetry [S_{pO_2}]) was 93%. Table 1 shows the concurrent arterial blood gas and hemoximetry data.

Question

What are the possible causes for the discrepancy between S_{pO_2} and the oxygen saturation measured via hemoximetry (oxyhemoglobin) in this case?

Answers

1. Spuriously high S_{pO_2} value
2. Inadvertent venous blood sampling
3. Excessive time delay in sample analysis
4. Consumption of oxygen by leukocytes in the blood sample ("leukocyte larceny")

Discussion

It is unlikely that an inaccurate S_{pO_2} reading explains the disparity between the S_{pO_2} measurement and the oxyhe-

moglobin value in this case. A common cause of spuriously high S_{pO_2} readings is the presence of above-normal carboxyhemoglobin,¹ but carboxyhemoglobin was only 0.6% (see Table 1). Other factors that can create spurious S_{pO_2} readings,^{2,3} such as intravascular dyes, dark skin pigmentation, motion artifact, nail polish, optical shunting, and poor peripheral perfusion, were not present. Also, recent work by Fluck et al⁴ refutes the idea that ambient light sources significantly affect pulse oximeter performance.

Inadvertent venous blood sampling is certainly a plausible explanation for the S_{pO_2} -oxyhemoglobin disparity in this case. It is recommended that pulse oximetry be used to help determine whether a hypoxemic blood specimen is in fact arterial, if there is clinical doubt. Inadvertent venous blood sampling is an important pre-analytic error against which all blood gas phlebotomists must be vigilant. Clearly, unrecognized inadvertent venous sampling can subject the patient to substantial morbidity and expenses from unnecessary interventions. Obviously, this does not occur only with pure venous samples; any amount of venous blood in an arterial sample results in a false reading. Therefore the ideal arterial puncture site is devoid of large adjacent veins.

Table 1. Arterial Blood Gas Values

pH	7.39
P_{aCO_2} (mm Hg)	43.6
P_{aO_2} (mm Hg)	49.5
HCO_3^- (mmol/L)	25.9
BE (mmol/L)	1.3
Hb (g/dL)	9.8
O_2Hb (%)	83.6
COHb (%)	0.6
MetHb (%)	0.1
RHb (%)	15.7
C_{aO_2} (vol %)	11.5

HCO_3^- = bicarbonate
 BE = base excess
 Hb = hemoglobin
 O_2Hb = oxyhemoglobin
 COHb = carboxyhemoglobin
 MetHb = methemoglobin
 RHb = reduced hemoglobin
 C_{aO_2} = oxygen content in arterial blood

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Ultimately, the S_{pO_2} -oxyhemoglobin disparity in this case was judged not to be due to inadvertent venous sampling. The respiratory therapist who drew the sample was very experienced and had high confidence that the sample was arterial because of the brisk, pulsatile filling of the syringe with blood. In addition, subsequent blood gas samples generated the same S_{pO_2} -oxyhemoglobin disparity.

Another pre-analytic error that can cause S_{pO_2} -oxyhemoglobin disparity is delay of blood sample analysis. Though erythrocytes derive much of their energy from anaerobic sources, other blood cells can consume considerable amounts of oxygen in vitro, which can lower the measured oxygen tension and content.

It is recommended that blood gas samples stored at room temperature be analyzed within 30 min of procurement.⁵ Others have recommended that samples should be placed in ice to reduce oxygen consumption if the specimen cannot be analyzed within 10 min of procurement.⁶ That type of pre-analytic procedural error did not occur in this case. The blood gas sample was analyzed within 3 min of procurement.

Ultimately, it was concluded that the cause of the S_{pO_2} -oxyhemoglobin disparity in this case was from oxygen consumption by leukocytes in the blood sample—a phenomenon known as “leukocyte larceny”; between the time the sample is drawn and the time the blood is analyzed, aerobic cells consume oxygen in vitro, thereby lowering the oxyhemoglobin saturation measurement. Extreme leukocytosis (886,000 cells/ μ L in this case) can be associated with rapid in vitro oxygen consumption, even if the sample is immediately cooled and analyzed within 10 min.⁷ Fox et al⁸ demonstrated 2-min declines in oxygen tension, ranging from 13 to 72 mm Hg, in leukemia patients, with leukocyte counts ranging from 55,000 cells/ μ L to 490,000 cells/ μ L. Placing a sample on ice and adding potassium cyanide can be effective in halting the consumption of oxygen by leukocytes.^{8,9} Another strategy is to use plasma to measure the oxygen tension.¹⁰

Though extreme leukocytosis can spuriously lower measured oxygen indices, one must not lose sight of the fact that leukocyte larceny and gas exchange dysfunction often coexist. Patients with leukemia are susceptible to respiratory dysfunction from chemotherapy agents, malignancy, and pneumonia. In addition, patients with lymphocytic leukemia and those who undergo bone marrow transplantation can develop pulmonary veno-occlusive disease.¹¹

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